## **MEMO/EXPECTED RESPONSES**

Grade 10 March 2018

| QUES               | STION 1: MULTIPLE CHOICE QUESTIONS   |            |
|--------------------|--|------------|
| 1.1                | D√√  | (2)        |
| 1.2                | A√√  | (2)        |
| 1.3                | B√√  | (2)        |
| 1.4                | A✓✓  | (2)        |
| 1.5                | B√√  | (2)        |
|                    |  | [ìÓ]       |
|                    | STION 2  |            |
| 2.1                | The temperature of a liquid at which its vapour pressure equals the external (atmosp | *          |
|                    | pressure. ✓ ✓  | (2)        |
| 2.2                | Nitrogen. ✓  | (1)        |
| 2.3                | Copper√  | (0)        |
|                    | Potassium chloride✓  | (2)        |
| 2.4                | Potassium chloride√  | (1)        |
| 2.5                | -78°C√   | <u>(1)</u> |
| OHES               | STION 3  | [7]        |
| <b>Q∪⊑3</b><br>3.1 | The name of the elements involved.✓  |            |
| J. I               | The <u>ratio</u> number of each element. ✓   | (2)        |
| 3.2.1              | CIO <sub>3</sub> ✓   | (2)        |
| 3.2.2              | $\operatorname{Cr_2O_7}^{2-}\checkmark$  | (2)        |
| 3.3.1              | sulphate ion ✓   | (-)        |
| 3.3.2              | permanganate ion ✓   | (2)        |
| 0.0.2              | pormanganato ion   | [06]       |
| QUES               | STION 4  |            |
| 4.1                | NaCl√  | (1)        |
| 4.2                | CO <sub>2</sub> ✓  | (1)        |
| 4.3                | Na√  | (1)        |
| 4.3                | SiO <sub>2</sub> ✓   | (1)        |
| 4.5                | Si√  | (1)        |
| 4.5.1              | Metalloids have properties of metals and non-metals. ✓✓                              | (2)        |
| 4.5.2              | Metalloids increase in conductivity with increasing temperature.✓                    |            |
|                    | Metals decrease in conductivity with increase in temperature. ✓                      | (2)        |
| 4.6                | $^{23}_{11}Na \checkmark\checkmark$  | (2)        |
|                    |  | (2)        |
| 4.7                | $1s^2 2s^2 2p^6 3s^1 \checkmark \checkmark$  | (2)        |
| 4.8                | •Na✓✓  | (2)        |
| OUES               | STION 5  | [15]       |
| <b>5</b> .1        | The temperature of a substance is a measure of the average kinetic energy            |            |
| J. I               | of the particles. $\checkmark\checkmark$   | (2)        |
| 5.2                | Thermometer ✓  | (2)        |
|                    | time√  | (1)        |
| 5.3.1<br>5.3.2     |  | (1)        |
|                    | temperature of the classroom.  | (1)        |
| 5.3.3<br>5.4       | temperature of the classroom√  | (1)        |
| 5.4<br>5.5         | Graph paper  | (5)        |
| 5.6                | -9 (°C) ✓ Constant temperature. ✓  | (1)        |
| J.U                | Latent heat is the energy absorbed <u>overcome the intermolecular forces</u> . ✓     | (2)        |
|                    | Eatent neat is the energy absorbed overcome the intermolecular forces.               | [14]       |
|                    |  | [, , ,     |

## **QUESTION 6**

- 6.1 Constructive interference is the phenomenon where the crest of one pulse overlaps with the crest of another to produce a pulse of increased amplitude. ✓✓ (2)
- Principle of superposition is the algebraic sum of the amplitudes of two pulses that occupy the same space at the same time.  $\checkmark\checkmark$  (2)

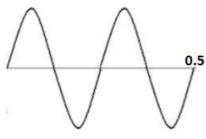
6.3 (5)

✓ Two crests or two troughs (any amplitude)
✓ Direction moving toward each other

✓ Larger amplitude.

✓ Two same sized crests or troughs.
✓ In the original directions

[09] **QUESTION 7** Transverse√ 7.1 (1) 7.2  $8m/3 = 2.67 \text{ m} \checkmark \checkmark$ (2) 7.3 Frequency: The number of wave pulses per second. (2)2 Hz√√ 7.4 (2)7.5 f x λ√ 2 x 2.67✓ 5.34 m·s<sup>-1</sup>√ (3)(2)7.6



✓ two waves shown✓ 0.5 shown

[12]

**TOTAL = 75** 

